

**BANGALORE SAHODAYA QUESTION PAPER (2022-23)**

**Science (Code – 086)**

**CLASS X –SET 2**

**Maximum Marks: 80**

**MARKING SCHEME**

**Time allowed: 3 Hrs**

Q.No	Questions	Marks
1.	d)	1
2.	c)	1
3.	b)	1
4.	b)	1
5.	d)	1
6.	c)	1
7.	c)	1
8.	b)	1
9.	b)	1
10.	c)	1
11.	b)	1
12.	b)	1
13.	b)	1
14.	c)	1
15.	c)	1
16.	d)	1
17.	b)	1
18.	b)	1
19.	d)	1
20.	a)	1
21.	<p>a) i) Anode and Cathode labeling must be switched.</p> <p>b) for Hydrogen: bring a burning splinter near the mouth of the test tube, it burns with “pop” sound.</p> <p>for oxygen: bring a burning splinter near the mouth of the test tube, it burns brightly</p> <p align="center"><b>OR</b></p> <p>a) <math>\text{BaCl}_2(\text{aq}) + \text{Na}_2\text{SO}_4(\text{aq}) \rightarrow \text{BaSO}_4 (\text{ppt}) + 2\text{NaCl}(\text{aq})</math></p> <p>b) <math>2\text{AgCl}(\text{s}) \xrightarrow{\text{sunlight}} 2 \text{Ag}(\text{s}) + \text{Cl}_2(\text{g})</math></p>	<p>1</p> <p>½</p> <p>½</p> <p>1</p> <p>1</p>
22.	<p>a. CNS- brain and spinal cord      PNS- Cranial and spinal nerves</p> <p>b. CNS- Processing of messages      PNS- Transmission of nerve impulses</p>	<p>1</p> <p>1</p>

23.	<p>a. Receptors receive the stimuli Gustatory receptor</p> <p>b. Due to presence of nervous and endocrine system/electrochemical and chemical coordination</p>	<p>½</p> <p>½</p> <p>1</p>
24.	<p>a. Stomatal pores open due to increase in turgor pressure. Stomatal pores close due to loss of turgor pressure.</p> <p>b. chloroplast photosynthesis</p>	<p>½</p> <p>½</p> <p>½</p> <p>½</p>
25.	<p>Fig 13.6 (a), Page 222 NCERT</p> <p>(i) Current in the conductor</p> <p>(ii) Distance from the conductor</p> <p>OR</p> <p>Upwards / out of the paper</p> <p>Statement of Fleming's left hand rule.</p>	<p>1</p> <p>½</p> <p>½</p> <p>1</p> <p>1</p>
26.	<p>a. Only <u>10% of the energy</u> is transferred from one trophic level to the next hence very less energy is available at the higher trophic levels</p> <p>b. High energy UV rays split molecular oxygen into free oxygen atoms. These free oxygen atoms combine with molecular oxygen to form ozone /</p> $\text{O}_2 \xrightarrow{\text{UV}} \text{O} + \text{O}$ $\text{O} + \text{O}_2 \rightarrow \text{O}_3 \text{ (Ozone)}$	<p>1</p> <p>1</p>
27.	<p>a) The interionic force of attraction in CaO is very strong when compared to magnesium chloride.</p> <p>b)</p> <div style="text-align: center;"> <p style="text-align: center;"> <math>\text{Ca} : + \ddot{\text{O}} : \Rightarrow [\text{Ca}]^{2+} + [:\ddot{\text{O}}:]^{2-} \Rightarrow \text{CaO}</math> </p> <p style="text-align: center;"> <small>[2, 8, 8, 2]      [2, 6]</small>  <small>Calcium atom      Oxygen atom</small> </p> <p style="text-align: center;">Calcium oxide</p> </div> <p>Electron dot structure of atoms – ½ each. ions ½ each</p>	<p>1</p> <p>2</p>
28.	<p>a) Calcium oxychloride and <math>\text{CaOCl}_2</math></p> <p>b) <math>\text{Ca}(\text{OH})_2 + \text{Cl}_2 \rightarrow \text{CaOCl}_2 + \text{H}_2\text{O}</math></p> <p>c) Bleaching powder is used in the paper industry to bleach wood pulp / disinfectant.</p> <p>d) It is used as an oxidising agent in the chemical industry</p>	<p>½ + ½</p> <p>1</p> <p>½</p> <p>½</p>
29.	<p>a) A- Glomerulus Function- Ultrafiltration/ filtration of blood</p> <p>b) The structure is coiled to increase the surface area for selective reabsorption</p>	<p>½</p> <p>½</p>



	<p>(iii) Ethyl ethanoate</p> <p>c) The above experiment must be carried out in a water bath as both compounds are volatile.</p> <p>d) used cleaning agent/ salad dressing/preservative. ( any relevant use)</p> <p style="text-align: center;"><b>OR</b></p> <p>a. A <math>\begin{array}{c} \text{H} \quad \text{H} \\   \quad   \\ \text{H}-\text{C}-\text{C}-\text{O}-\text{H} \\   \quad   \\ \text{H} \quad \text{H} \end{array}</math> B. <math>\text{CH}_3-\overset{\text{O}}{\parallel}{\text{C}}-\text{H}</math></p> <p>b. A is ethanol and B is ethanal.</p> <p>c. A is alcohol / OH and B is aldehyde / CHO</p> <p>d. <math>\text{CH}_3-\text{CH}_2-\text{OH} \xrightarrow[\text{Or Acidified } \text{K}_2\text{Cr}_2\text{O}_7]{\text{Alkaline } \text{KMnO}_4} \text{CH}_3\text{COOH}</math></p> <p>e) (i) antifreeze agent / Tincture of iodine. (ii) used in cough syrups (any relevant use can be written)</p>	<p>1</p> <p>1</p> <p><math>\frac{1}{2} + \frac{1}{2}</math></p> <p><math>\frac{1}{2} + \frac{1}{2}</math></p> <p><math>\frac{1}{2} + \frac{1}{2}</math></p> <p>1</p> <p><math>\frac{1}{2} + \frac{1}{2}</math></p>
35.	<p>a. Asexual reproduction/binary fission</p> <p>Example- Bacteria / amoeba / Leishmania / paramecium / euglena</p> <p>b. Advantage- Results in large number of offspring at a rapid rate</p> <p>Disadvantage- Less variation</p> <p>c. Due to errors in DNA replication</p> <p>d. Variations help the organism to adapt to changing ecological niche.</p> <p style="text-align: center;"><b>OR</b></p> <p>a. Fertilisation takes place in the oviduct/fallopian tube</p> <p>Implantation takes place in the uterus</p> <p>b. i. The endometrial lining thickens and is richly supplied with blood to nourish the growing embryo</p> <p>ii. The lining slowly breaks and comes out through the vagina as blood and mucous.</p> <p>c. Causes irritation of the uterus</p> <p>Does not prevent transmission of sexually transmitted diseases</p> <p>d. To prevent adverse effect on the health of the mother/ to give time to mature mentally/ any other correct reason</p>	<p><math>\frac{1}{2}</math></p> <p><math>\frac{1}{2}</math></p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p><math>\frac{1}{2}</math></p> <p><math>\frac{1}{2}</math></p> <p>1</p> <p>1</p> <p><math>\frac{1}{2}</math></p> <p><math>\frac{1}{2}</math></p> <p>1</p>
36.	<p>(a) Tungsten has high melting point and incandescence</p> <p>(b) Ammeter measures current which is the same in a series circuit..</p>	<p>1</p> <p>1</p>

	<p>(c) <math>6\Omega</math> II <math>3\Omega</math>, <math>R_p = 2\Omega</math></p> <p><math>R_{\text{eff}} = 2 + 2 + 5 = 9\Omega</math></p> <p><math>I = V/R = 12/9 = 4/3 \text{ A} / 1.33\text{A} / 3.56\text{W}</math></p> <p><math>P = I^2R = 4/3 \times 4/3 \times 2 = 32/9 \text{ W}</math></p>	<p><math>1\frac{1}{2}</math></p> <p><math>\frac{1}{2}</math></p> <p>1</p>
37.	<p>a. both are oxidising agents</p> <p>b. <math>2\text{Mg} + \text{SO}_2 \text{ -----} &gt; 2\text{MgO} + \text{S}</math></p> <p>c. <math>\text{SO}_2 \longrightarrow</math> acidic in nature.</p> <p>Blue litmus turns red in aq <math>\text{SO}_2</math> / wet blue litmus turns red in <math>\text{SO}_2</math></p> <p style="text-align: center;"><b>OR</b></p> <p><math>\text{MgO} \longrightarrow</math> basic in nature</p> <p>Red litmus turns blue in aq <math>\text{MgO}</math>.</p>	<p>1</p> <p>1</p> <p>1</p> <p><math>\frac{1}{2} + \frac{1}{2}</math></p> <p>1</p> <p><math>\frac{1}{2} + \frac{1}{2}</math></p>
38.	<p>a. Genes are located in chromosome/chromatin thread/ DNA in the nucleus of pea plant. (only 0.5 marks to be given if only nucleus is mentioned)</p> <p>b. Purple is the dominant trait and white flowers is the recessive trait. Reason- White flowers are produced when a non-efficient protein is coded by the allele 'a'. Therefore, white is the recessive trait.</p> <p>c. 1 (purple): 1 (white)</p> <p style="text-align: center;"><b>OR</b></p> <p>c. Phenotype- all purple flowers Genotype- AA, Aa</p>	<p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p><math>\frac{1}{2}</math></p> <p><math>\frac{1}{2}</math></p>
39.	<p>(a) Scattering of light.</p> <p>(b) No atmosphere to scatter light.</p> <p>(c) Smaller particles scatter smaller wavelengths and blue has a smaller wavelength. Hence it is scattered the most.</p> <p style="text-align: center;"><b>OR</b></p> <p>Red has the longest wavelength. Hence it is scattered the least.</p>	<p>1</p> <p>1</p> <p>2</p>

\*\*\* Good Luck\*\*\*